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Key indicators

Single-crystal X-ray study T = 120 KMean $\sigma(\text{C}-\text{C}) = 0.002 \text{ Å}$ R factor = 0.035 wR factor = 0.090Data-to-parameter ratio = 11.2

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

2,3,4,5,6-Pentafluorobenzamide

The dihedral angle between the amide group and the pentafluorophenyl moiety of the title compound, $C_7H_2F_5ON$, is 43.56 (5)°. The molecules are connected *via* N-H···O=C bonds into an infinite, two-dimensional network generated by the cyclic $R_2^2(8)$ and chain C(4) motifs.

Comment

A common pattern of assembly of primary amides is via centrosymmetric, cyclic $R_2^2(8)$ hydrogen-bond motif involving the syn-oriented NH_s groups, and via the C(4) motif with the use of anti-oriented NHa groups. Two situations may arise depending on the symmetry operation relating the amide molecules within the C(4) chain. When amide cyclic dimers are related by ca 5.1 Å translation, a one-dimensional assembly of hydrogen-bonded molecules is generated. In turn, the glide plane or 2_1 symmetry operating between the dimers leads to a two-dimensional molecular network (Leiserowitz & Hagler, 1983; Bernstein et al., 1994). The former assembly mode is generally preferred over the latter and is found, for example, in benzamide (Blake & Small, 1972) and in its 2- and 3-fluoro derivatives (Taniguchi et al., 1975; Kato & Sakurai, 1982), while molecules of 4-fluorobenzamide are hydrogenbonded to form the two-dimensional network (Kubota & Ohba, 1992). On the other hand, 4-chlorobenzamide, which crystallizes in three polymorphic forms, can create both types of assemblies (Taniguchi et al., 1978; Hayashi et al., 1980). Whereas the crystal structure of 2,3,4,5,6-pentafluorobenzoic acid has been known for over 30 years (Benghiat & Leiserowitz, 1972), the structure of its amide analogue, (I), has not been reported so far.



The structure of (I), with the atom-numbering scheme, is shown in Fig. 1. The molecule of (I) is not planar and its amide group is *ca* 15° more twisted relative to the aromatic ring than in the benzamide structure (Blake & Small, 1972). The largest difference in the geometry of the two amide molecules occurs in the value of the endocyclic bond angle at C1, which is 120.1 (3)° in benzamide and only 116.59 (13)° in its fluoro analogue. Similar changes have been observed in the geometries of benzoic (Bruno & Randaccio, 1980) and pentafluorobenzoic acids (Benghiat & Leiserowitz, 1972)

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The hydrogen-bonding geometry is given in Table 2. In contrast to benzamide, the molecules of (I) are assembled in the crystal structure into a two-dimensional hydrogen-bonded network parallel to the (001) plane (Fig. 2). The amide NH_a groups take part in hydrogen bonds joining glide-planerelated centrosymmetric $R_2^2(8)$ dimers. Each molecule gives rise to five short intermolecular $F \cdots F$ contacts in the range 2.8117 (13)–2.8605 (13) Å. No significant intermolecular $\pi - \pi$ stacking interactions are found between the perfluorophenyl rings; however, there are two short $F \cdot \cdot \cdot C$ contacts of 2.927 (2) and 3.016 (2) Å. They occur between the meta-fluoro group F5 and the benzene C3 and carbonyl C7 atoms of molecules related by a unit translation along b. As has been shown recently, the $C-F \cdots O = C$ interactions are robust in fluorinecontaining organic carbonyl compounds. They are best described as multipolar interactions between the intrinsically polar C-F and C=O groups (Olsen et al., 2003).

Experimental

Commercially available 2,3,4,5,6-pentafluorobenzamide was recrystallized by slow evaporation of an ethanol solution.



Figure 1

View of the molecule of (I), shown with 50% probability displacement ellipsoids.



Figure 2

Packing diagram for (I), viewed along the b axis. Hydrogen bonds and short $F \cdots F$ contacts are shown by dashed lines.

Crystal	data
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C ₇ H ₂ F ₅ NO	$D_{\rm r} = 1.884 {\rm Mg m}^{-3}$
$M_r = 211.10$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/c$	Cell parameters from 2021
a = 12.7305 (11) Å	reflections
b = 6.1929 (6) Å	$\theta = 4-25^{\circ}$
c = 9.5856 (10) Å	$\mu = 0.21 \text{ mm}^{-1}$
$\beta = 100.073 (8)^{\circ}$	T = 120 (2) K
V = 744.07 (12) Å ³	Prism, colorless
Z = 4	$0.40 \times 0.20 \times 0.20 \text{ mm}$
Data collection	
Kuma KM-4 CCD diffractometer	$R_{\rm int} = 0.016$
ω scans	$\theta_{\rm max} = 26.4^{\circ}$
4010 measured reflections	$h = -15 \rightarrow 15$
1517 independent reflections	$k = -7 \rightarrow 7$
1376 reflections with $I > 2\sigma(I)$	$l = -11 \rightarrow 5$
Refinement	
Refinement on F^2	$w = 1/[\sigma^2(F_a^2) + (0.048P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.035$	+ 0.2307P]
$wR(F^2) = 0.090$	where $P = (F_{a}^{2} + 2F_{c}^{2})/3$
S = 1.13	$(\Delta/\sigma)_{\rm max} = 0.002$
1517 reflections	$\Delta \rho_{\rm max} = 0.27 \ {\rm e} \ {\rm \AA}^{-3}$
135 parameters	$\Delta \rho_{\rm min} = -0.26 \ {\rm e} \ {\rm \AA}^{-3}$

Table 1

Selected geometric parameters (Å, °).

All H-atom parameters refined

O1-C7 N1-C7	1.2392 (17) 1.3217 (19)	C1-C7	1.513 (2)	
C2-C1-C6 O1-C7-N1	116.59 (13) 123.90 (14)	O1-C7-C1 N1-C7-C1	119.08 (13) 117.02 (12)	
C2-C1-C7-N1	44.8 (2)	C6-C1-C7-O1	41.2 (2)	

Table 2

Hydrogen-bonding geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdots A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$\begin{array}{c} N1 - H1 \cdots O1^{i} \\ N1 - H2 \cdots O1^{ii} \end{array}$	0.88 (2)	2.05 (2)	2.8902 (17)	158.3 (17)
	0.89 (2)	2.03 (2)	2.9137 (17)	175.5 (16)

Symmetry codes: (i) $x, \frac{3}{2} - y, z - \frac{1}{2}$; (ii) -x, 2 - y, -z.

Data collection: CrysAlisCCD (Oxford Diffraction, 2000); cell refinement: CrysAlisCCD; data reduction: CrysAlisRED (Oxford Diffraction, 2000); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: Stereochemical Workstation (Siemens, 1989); software used to prepare material for publication: SHELXL97.

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